L13: Controlled Drug Delivery Systems

May 17, 2018

Introduction & Definition

Controlled drug delivery systems offer an approach to regulating both the duration and spatial localization of therapeutic agents.

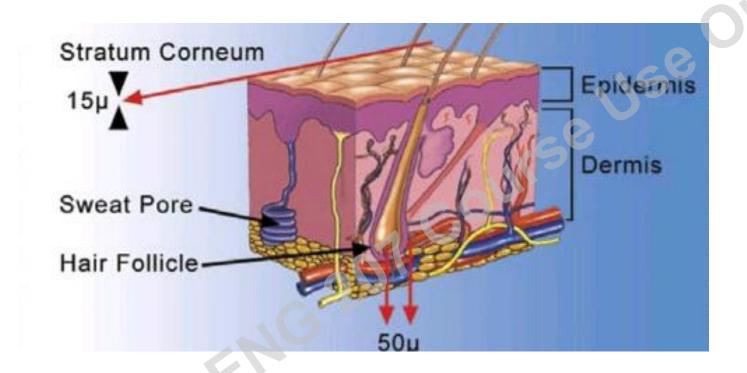
Controlled drug delivery systems:

- --- Include a component that can be engineered to regulate an essential characteristic (e.g., duration of release, rate of release, or targeting).
- --- Have a long duration of action (> 1 day)

Content

- ❖ Reservoir and transdermal delivery systems
- **❖** Matrix delivery systems
- Hydrogel delivery systems
- Degradable delivery systems
- Particulate delivery systems
- Responsive delivery systems

I. Reservoir and Transdermal Delivery Systems



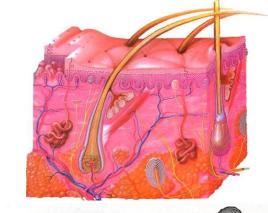
Schematic representation of a cross section through human skin.

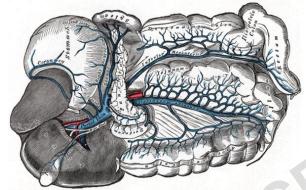
Stratum corneum, located on the outer surface of the skin, is a non-living layer of keratin-filled cells surrounded by a lipid-rich extracellular matrix that provides the primary barrier to drug delivery into skin.

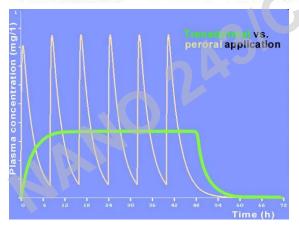
The **epidermis** below is a viable tissue devoid of blood vessels.

Just below the dermal-epidermal junction, the **dermis** contains capillary loops that can take up transdermally administered drugs for systemic distribution.

Advantages of Transdermal Delivery



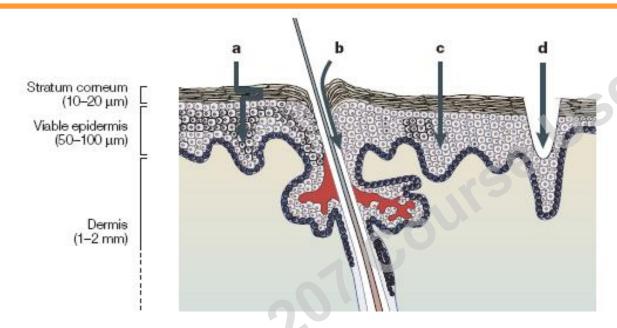




After transdermal application, the drug penetrates through the skin and reaches the systemic circulation. Transdermal systems are advantageous for several drugs because:

- The drug does not reach the gastrointestinal tract. Thus, it can not be inactivated or produce side effects in this area.
- The drug does not undergo first pass metabolism in the liver, therefore less drug needs to be delivered.
- Sustained and controlled plasma levels can be obtained, avoiding peak levels which might produce side effects.
- The patient compliance is improved. The patient can easily control whether or not to take the medicine.
- The patient can immediately stop the administration by removing the patch.

Current Status of Transdermal Delivery



- (a) Transdermal diffusion, possibly in the presence of a chemical enhancer, takes place by a tortuous route across the stratum corneum, winding around cells and occurring along the interfaces of extracellular lipid bilayers.
- (b) Low-voltage electrical enhancement by iontophoresis can make transport pathways through hair follicles and sweat ducts more accessible.
- (c) High-voltage enhancement by electroporation has been shown to occur via transcellular pathways made accessible by disrupting lipid bilayers.
- (d) Microneedles and thermal poration create micron-scale holes in skin to provide pathways for drug transport.

Transdermal Patches



- (1) Liner Protects the patch during storage.
- (2) Drug Drug solution in direct contact with release liner.
- (3) Adhesive Serves to adhere the components of the patch together along with adhering the patch to the skin.
- (4) Membrane Controls the release of the drug from the reservoir and multi-layer patches.
- (5) Backing Protects the patch from the outer environment.

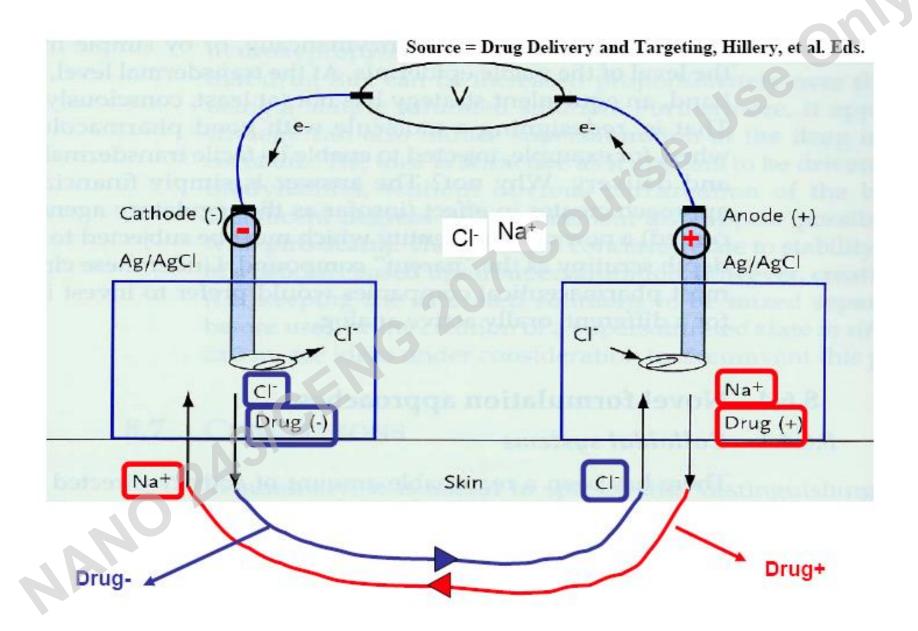




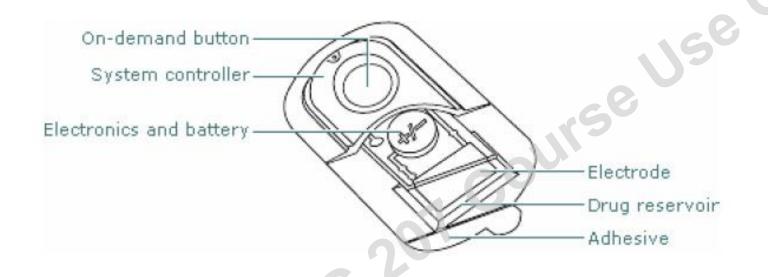
CombiPatch: A transdermal patch marketed by Novogyne Pharmaceuticals for hormone replacement therapy.

Nicotinell: A transdermal patch marketed by Novartis Pharmacological for smoking cessation

Transdermal Iontophoretic Drug Delivery

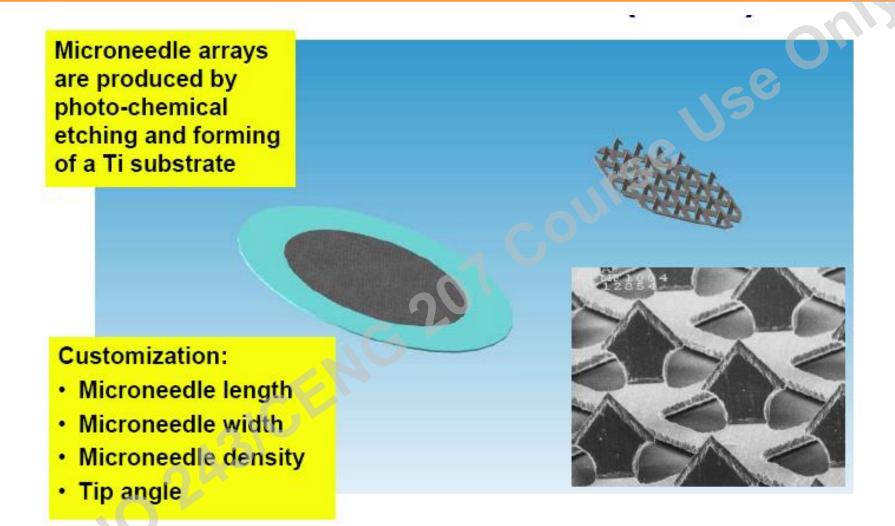


Transdermal Iontophoretic Drug Delivery



ALZA Corporation (Mountain View, CA), a division of Johnson and Johnson, has been developing an iontophoresis product, called E-TRANS®, for local or systemic administration of drug compounds. According to ALZA, design options include disposable systems for delivering drug over hours or days, and reusable systems with replaceable batteries and drug pads. The first product is designed to alleviate acute post-operative pain with patient-administered doses of fentanyl, a potent opioid analgesic.

Transdermal Microneedles



Macroflux® Patch (Alza)

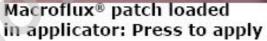
Transdermal Microneedles

Drug-coated Macroflux® patch







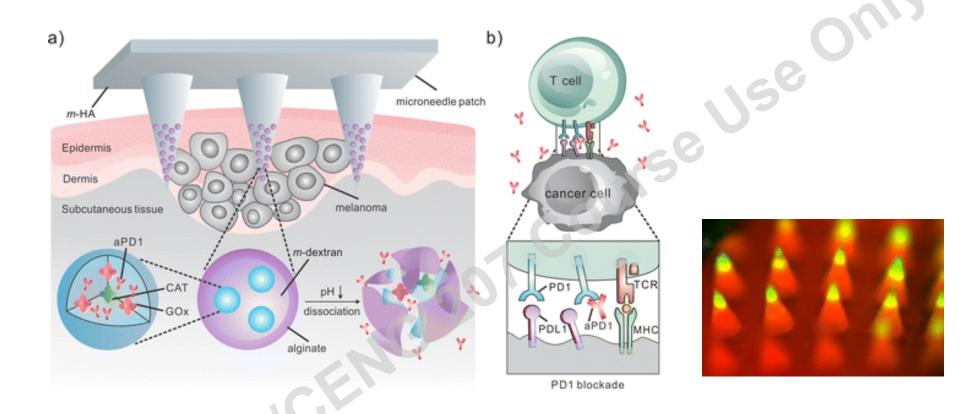




Post application

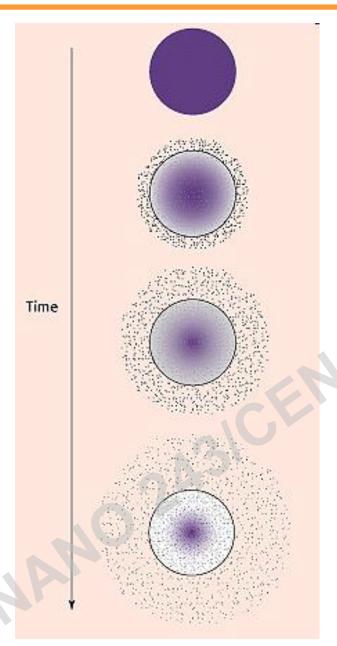
Patch Application is Controlled by a Spring-Loaded Applicator System (Alza)

Recent Development of Microneedles



Schematic of the MN patch-assisted delivery of aPD1 for the skin cancer treatment. (a) Schematic of the aPD1 delivered by an MN patch loaded with physiologically self-dissociated NPs. With GOx/CAT enzymatic system immobilized inside the NPs by double-emulsion method, the enzyme-mediated conversion of blood glucose to gluconic acid promotes the sustained dissociation of NPs, subsequently leading to the release of aPD1. (b) The blockade of PD-1 by aPD1 to activate the immune system to destroy skin cancer cells.

II. Matrix Delivery Systems



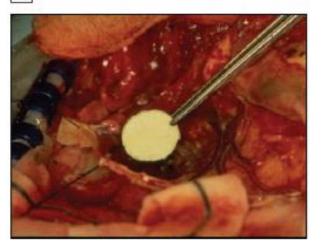
For some therapeutic agents, it is not possible to find membrane materials that provide adequate permeability to permit release from a reservoir device.

In a matrix system, the drug molecules are dissolved or dispersed throughout a solid polymer phase.

The drug molecules will continuously diffuse out of the matrix

Biodegradable Polymer Matrix

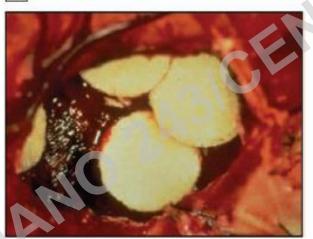
a



Biodegradable polymers that dissolve slowly have been widely used in matrix drug delivery systems.

By careful design of the material and device, it is possible to design delivery systems in which the rate of polymer degradation and dissolution controls the rate of drug delivery, providing a new element for controlling the rate of release of dispersed drugs.

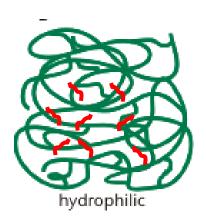
b



Polymer-based local chemotherapy. (a) Photograph of a polymer wafer containing chemotherapeutic drug BCNU being surgically inserted into a human brain. (b) Seven or eight wafers are inserted in the surgical site before the surgeon closes the brain.

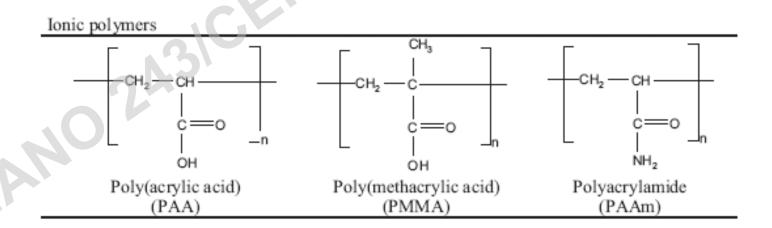
III. Hydrogel Delivery Systems

Water-soluble polymers can be cross-linked to create materials, called hydrogel, that swell but do not dissolve in water.



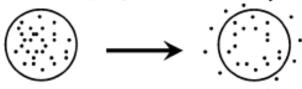
Cross-links (covalent bonds or physical entanglements) convert the ensemble of individual polymer chains into a macromolecular network, to which drugs are encapsulated.

The rate of drug diffusion through the hydrogels depends on the extent of cross-linking (interchain separation).



Mechanisms of Hydrogel Drug Delivery

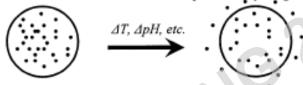
Drug dissolved in hydrogel



Drug encapsulated in hydrogel



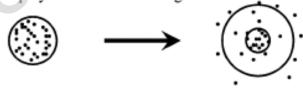
Drug dissolved in enviro-stimuli-responsive hydrogel



Drug in degradable hydrogel



Swollen polymer from which drug is released



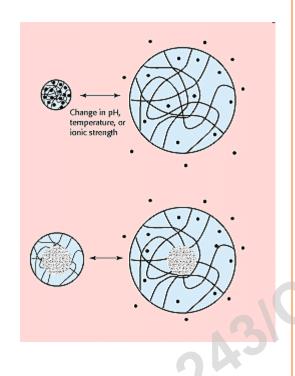
Time 0

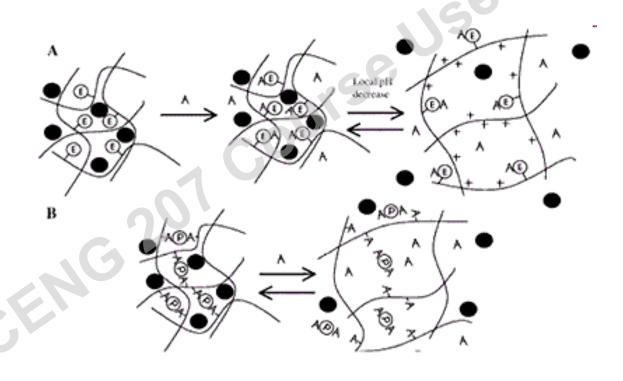
Time t

Three primary mechanisms:

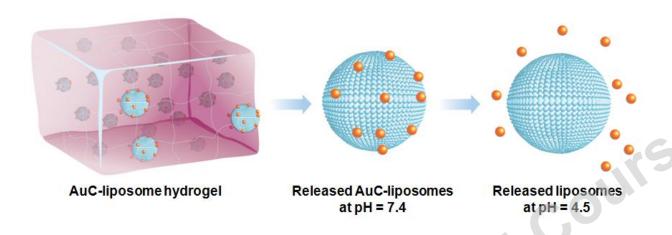
- Diffusion
- Degradation
- Swelling followed by diffusion

Swelling Followed by Diffusion

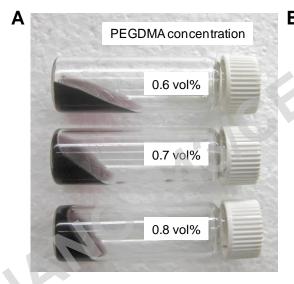


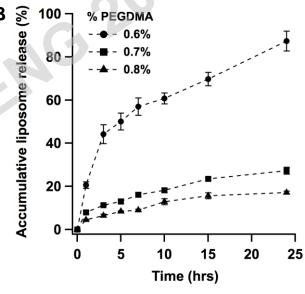


Hydrogel for Topical Antimicroibal Delivery



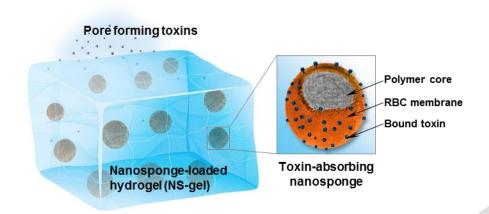
Schematic illustration of hydrogel containing nanoparticle-stabilized liposomes for topical antimicrobial delivery



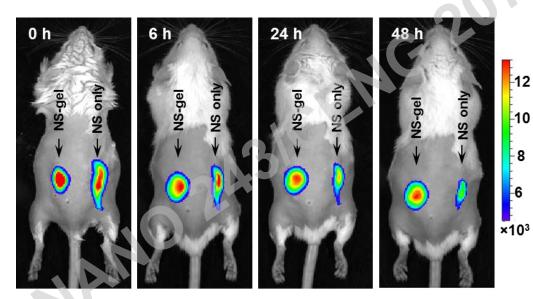


Formulation and drug release rate of AuC-liposome hydrogels.

Hydrogel-Nanoparticle Hybrid System



Schematic illustration of a hydrogel retaining toxin-absorbing nanosponges for local treatment of methicillin-resistant
Staphylococcus aureus (MRSA) infection.
The toxin nanosponge was constructed with a polymeric core wrapped in natural red blood cell (RBC) bilayer membrane and was subsequently embedded into an acrylamide-based hydrogel.



In vivo nanosponge retention by hydrogel. Nanosponges labeled with DiD fluorescent dye was used to formulate NS-gel, which was then injected subcutaneously under the loose skin over the left flank of the mice. Free suspended nanosponges (without hydrogel) were injected as a control group at the right flank of the same mice. Fluorescence images taken at different time points show the retention of the nanosponges under mouse skin.

IV. Degradable Delivery Systems

Biodegradable polymers degrade within the body as a result of natural biological processes, eliminating the need to remove a drug delivery system after release of the active agent has been completed.

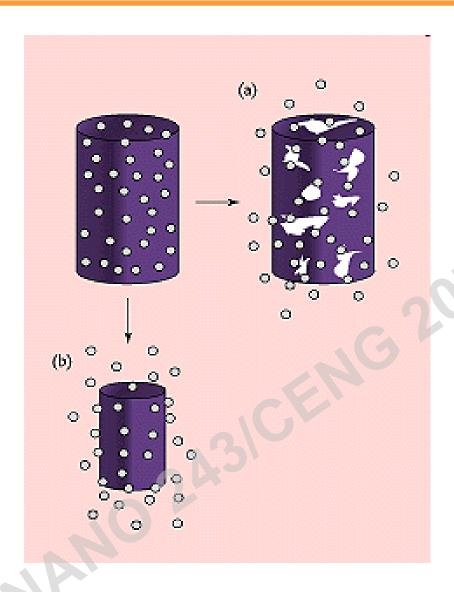
Key requirements:

- (a) The biomaterials degrade into naturally occurring or inert chemicals, which are not harmful to the body.
- (b) The biomaterials degrade in a controllable fashion.

Most biodegradable polymers are designed to degrade as a result of hydrolysis of the polymer chains into biologically acceptable, and progressively smaller, compounds.

In some cases—as, for example, polylactides, polyglycolides, and their copolymers—the polymers will eventually break down to lactic acid and glycolic acid, enter the Kreb's cycle, and be further broken down into carbon dioxide and water and excreted through normal processes.

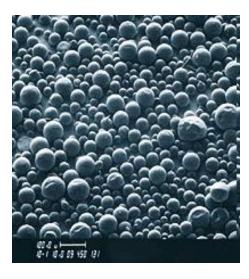
Bulk Erosion vs. Surface Erosion



Bulk erosion: Degradation may take place through bulk hydrolysis, in which the polymer degrades in a fairly uniform manner throughout the matrix, as shown schematically in Figure a.

Surface erosion: For some degradable polymers, most notably the polyanhydrides and polyorthoesters, the degradation occurs only at the surface of the polymer, resulting in a release rate that is proportional to the surface area of the drug delivery system, as shown schematically in Figure b.

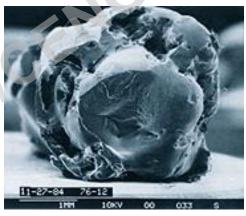
Bulk Erosion vs. Surface Erosion





Bulk erosion: Biodegradable microparticle of 75:25 lactide:glycolide PLGA after 133 days of degradation in water

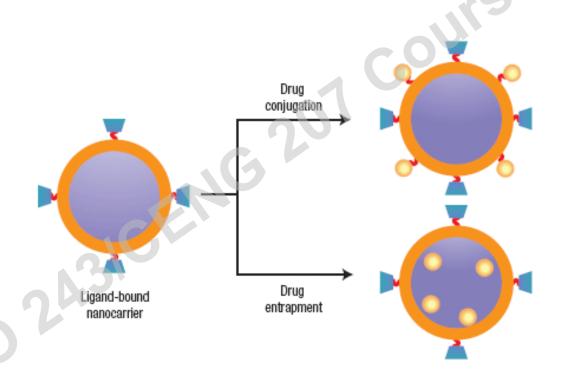




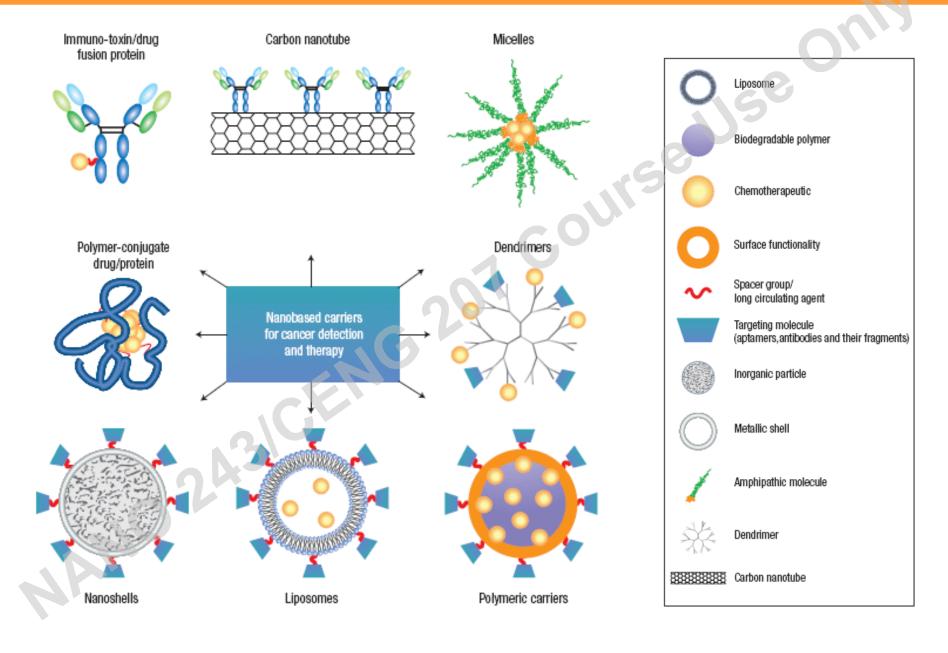
Surface erosion: Biodegradable polyorthoester rods after (left) 9 and (right) 16 weeks of implantation in rabbits.

V. Particulate Delivery Systems

While implantable drug delivery systems are useful for certain applications, in many cases an injectable or ingestable delivery system is desired. For that reason, particulate delivery systems represent an important class of drug delivery systems.



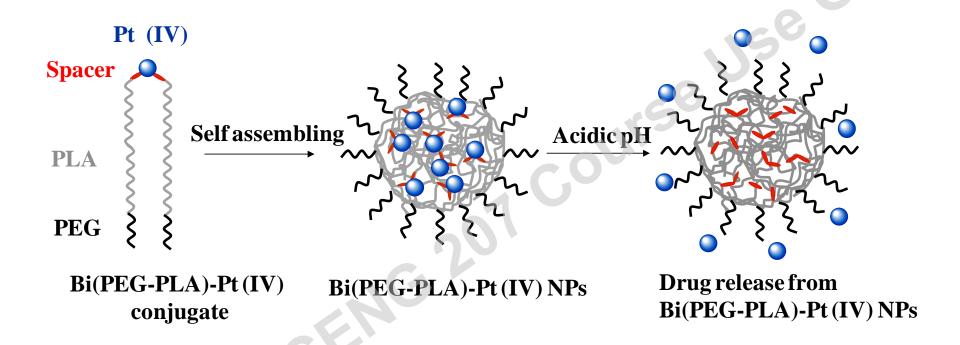
Typical Particulate Delivery Systems



VI. Responsive Delivery Systems

Stimulus	Hydrogel	Mechanism
рН	Acidic or basic hydrogel	Change in pH — swelling — release of drug
Ionic strength	Ionic hydrogel	Change in ionic strength — change in concentration of ions inside gel — change in swelling — release of drug
Chemical species	Hydrogel containing electron-accepting groups	Electron-donating compounds — formation of charge/transfer complex — change in swelling — release of drug
Enzyme-substrate	Hydrogel containing immobilized enzymes	Substrate present — enzymatic conversion — product changes swelling of gel — release of drug
Magnetic	Magnetic particles dispersed in alginate microshperes	Applied magnetic field — change in pores in gel — change in swelling — release of drug
Thermal	Thermoresponsive hrydrogel poly(N-isopropylacrylamide)	Change in temperature — change in polymer- polymer and water-polymer interactions — change in swelling — release of drug
Electrical	Polyelectrolyte hydrogel	Applied electric field — membrane charging — electrophoresis of charged drug — change in swelling — release of drug
Ultrasound irradiation	Ethylene-vinyl alcohol hydrogel	Ultrasound irradiation — temperature increase — release of drug

pH-Triggered Drug Release



Schematic illustrations of the structure of polymer-cisplatin prodrug conjugate, the formation of Bi(PEG-PLA)-Pt(IV) NPs through self-assembling, and acid-responsive drug release from the NPs

Magnetically Controlled Drug Release

